\_\_\_\_\_ Date: \_\_

## Name: \_\_\_\_\_ Class: \_\_\_\_\_ Class:

1. Fill in the blanks below to show the differences between mass and weight.

	Mass	Weight
(a)		Is the force of gravity on the body
(b)	Is constant at any location	
(c)	Is a scalar quantity	
(d)	Is measured in kilograms	
(e)	Is measured using a beam balance	

- 2. Write down the relationship between mass m, weight W and acceleration due to gravity g.
- Acceleration due to gravity on the surfaces of the Earth and the Moon are 10 m s<sup>-2</sup> and 1.6 m s<sup>-2</sup> respectively. Complete the following table by using the information given.

	We	ight
Mass	on Earth	on Moon
1 kg		
100 g		
	600 N	
		8 N

- **4.** A ball of mass 0.5 kg was thrown vertically upwards into the air. It reached a maximum height of 5 m. [Take  $g = 10 \text{ N kg}^{-1}$ ]
  - (a) What was the magnitude of the force acting on the ball at the highest point?
  - (b) What was the acceleration of the ball at the highest point?

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(c) After the ball had fallen 1 m from its highest point, its acceleration decreased. Why?

5.	Wh	nat is meant by the term 'inertia'?	
6.	Cor	mplete these sentences.	
	(a)	Objects having greater mass will have inerti	а.
	(b)	The greater the mass of a body, the moreto stop it.	it is to start it moving or
	(C)	The law of inertia is consistent with Newton's	law of motion.
7.	Wh	at is meant by 'gravitational field'?	
8.	Def	ine gravitational field strength.	
9.		nplete these sentences.	
	(a)	The SI unit for gravitational field strength is the	
	(b) -	The gravitational field strength has the same value as the	due to gravity.
10.	roop	gravitational field strength of Earth, Jupiter and the Sun are 10 N kg <sup>-1</sup> , 2 pectively. The weight of an object on the surface of Jupiter is 130 N. What is the mass of the object?	$^{\circ}6$ N kg <sup>-1</sup> and 280 N kg <sup>-1</sup>

- (b) What is the weight of the object
  - (i) on the surface of Earth?

(ii) on the surface of the Sun?

Name:	Class:	Date:	
Exercise 4B Density			
1. (a) Define density.			

(b) What is the SI unit for density?

2. The density of water is 1 g cm<sup>-3</sup>. What is its density expressed in kg m<sup>-3</sup>?

3. Complete the following table to find either the mass, volume or density of the material.

		Volume	Density		
Material	Mass		g cm <sup>-3</sup>	kg m-³	
aluminium	135 g	50 cm <sup>3</sup>			
copper	178 g		8.9 g cm <sup>-3</sup>	8900 kg m <sup>-3</sup>	
glass		40 cm <sup>3</sup>	2.5 g cm <sup>-3</sup>	2500 kg m <sup>-3</sup>	
petrol	160 g	200 cm <sup>3</sup>			
mercury		5 cm <sup>3</sup>	13.6 g cm <sup>-3</sup>	13 600 kg m <sup>-3</sup>	

A rectangular block of metal measures 50 mm by 45 mm by 4 mm. Its mass is 24 g. Calculate
(a) the volume of the metal block,

(b) the density of the metal block in g cm<sup>-3</sup>.

5. A student was asked to check whether a gold chain is made of pure gold. He found that the gold chain had a mass of 300 g. Using the displacement method, the volume of the gold chain was measured to be 20 cm<sup>3</sup>. Could the gold chain be pure gold? (The density of gold is 19.3 g cm<sup>-3</sup>.)

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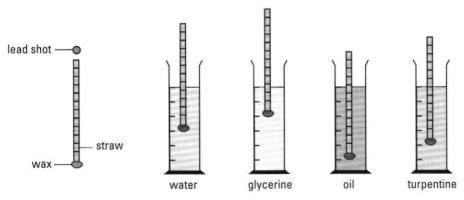
- 6. Four litres of petrol have a mass of 3.2 kg.
  - (a) What is the volume of the petrol expressed in terms of m<sup>3</sup>?
  - (b) Calculate the density of petrol.
- A classroom measures 5.0 m by 6.0 m by 3.0 m. The air inside it has a density of 1.3 kg m<sup>-3</sup>. Calculate
   (a) the volume of air in the classroom,
  - (b) the mass of the air.
- 200 cm<sup>3</sup> of water with density 1.0 g cm<sup>-3</sup> are mixed with 300 cm<sup>3</sup> of methylated spirit with density 0.80 g cm<sup>-3</sup>. Assuming there is no change in total volume after mixing, calculate
  - (a) the mass of water,
  - (b) the mass of methylated spirit,
  - (c) the total mass of the mixture,
  - (d) the total volume of the mixture,
  - (e) the density of the mixture.

Name:		Class: Date:					
$_{\text{Exercise}} 4C$ Floating and Sinking							
1.	Cor	nplete these sentences.					
	(a)	When an object is placed in a liquid of a lower density, it					
	(b)	When an object is placed in a liquid of a higher density, it					
	(c)	When an object is placed in a liquid having the same density as the object, it					

2. Three objects of densities 0.9 g cm<sup>-3</sup>, 1.0 g cm<sup>-3</sup> and 1.1 g cm<sup>-3</sup> respectively are placed in water of density 1.0 g cm<sup>-3</sup>. Draw in the diagram below to show how the three objects will settle in water.

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		and the second second	and the second s	— water (density 1.0 g cm

- **3.** When an object floats in various types of liquids, the denser the liquid, the \_\_\_\_\_\_. the object will float in the liquid.
- 4. The drawing below shows the relative positions of straws loaded with lead shots and placed in various types of liquids.



- (a) Which liquid has the highest density?
- (b) Which liquid has the lowest density?

- 9. An alloy is made by mixing 450 g of cobalt (density 9 g cm<sup>-3</sup>) with 240 g of iron (density 8 g cm<sup>-3</sup>).
  - (a) What is the volume of cobalt used?
  - (b) What is the volume of iron used?

(c) What is the total volume of cobalt and iron used?

(d) What is the density of the alloy?

10. An alloy is made by mixing metal A and metal B, which have densities of 12 g cm<sup>-3</sup> and 4 g cm<sup>-3</sup> respectively. The alloy weighs 100 g and is found to have a volume of 10 cm<sup>3</sup>. Fill in the blanks and find the mass of metal A. [Hint: Let *m* be the mass of metal A.]

Material		Mass		Volume		Density
Metal A	(a)	т	(C)	8 <del></del>	_ cm <sup>3</sup>	12 g cm <sup>-3</sup>
Metal B	(b)		(d)	2. 1.	_ cm <sup>3</sup>	4 g cm <sup>-3</sup>